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NEAR SIMULTANEOUS MEASUREMENTS OF NO₂ AND NO₃ OVER TROPICS BY GROUND-BASED ABSORPTION SPECTROSCOPY

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Abstract

The present study concentrates on measurements of NO_2 and NO_3 . NO_2 has been measured during twilight period using zenith sky absorption spectrometric technique in the 436 to 448 nm region. NO_3 has been measured during night time using direct moon as a source of light in the 655 to 667 nm region. These measurements have been taken at low latitude station, Ahmedabad (23°N, 76°E), India for the past two years.

Introduction

NO group of species plays an important role in the chemical reactions that control the distribution of O in the stratosphere. The variabilities of NO and NO are known to some extent, those of NO₃ are in considerable darkness. First observation of NO₂ was taken by Noxon et al. (1978), who used a scanning spectrometer and moon as a light source above the Fritz Peak Observatory (40°N) in April 1978. The same spectral feature has been used to study stratospheric NO₃ by several authors (Sanders et al., 1987; Solomon et al., 1989a,b). However, these observations are not sufficient to draw a clear picture of the global NO₃ variability. A number of observations have been taken on the NO₂ content of the stratosphere by using ground-based, balloon-borne and space craft techniques (Syed and Harrison, 1980; Mckenzie et al. 1982; Mount et al., 1987; Pommereau et al., 1988; Johnston et al. 1992). But most of these observations are of mid- and high latitude zones. Satellite observations are available for low latitude zone; but those are not yet fully validated due to the lack of enough observations in this region. distribution and variability of this species in the low latitude northern hemisphere, however, are not properly known. We have, therefore, initiated measurements of NO₂ and NO₃ over Ahmedabad, a low latitude station. The measurements reported here will contribute significantly to the global data base of minor neutral constituents.

Experiment

The details of experimental set-up has been described in the companion paper(Lal & Chakrabarty).

Methodology

Using the property that NO, has strong absorption at 662 nm, and NO₂ has strong absorption at 439.5 nm, NO₂ and NO₃ column abundances have been measured. Since the concentration of NO₃ is very low during daytime, its abundance has been measured during night time only. The technique of NO₂ measurement has been described by Noxon (1979), Harrison (1979) and Syed and Harrison (1980). The analysis of NO₃ has been done by using matrix inversion technique. For NO, measurements, the background spectrum has been obtained at minimum possible solar zenith angle. Superficially, twilight and background spectra appear similar owing to the presence of strong Fraunhofer structure in the incident sunlight. By taking their ratio, the Fraunhofer structure is removed. After this is done, absorption by NO, becomes apparent owing to the much longer path during twilight period. ratio spectrum has been used for the calculation of NO₂ abundance using five point method as given by Harrison (1979).

For NO₃ measurements, direct solar spectrum at 80° solar zenith angle has been used as a background spectrum. All the observations have been taken at higher lunar zenith angle during night time. Besides NO₃ absorption, aerosol and Rayleigh scattering, and absorption due to H₂O and O₃ are also present. These have been removed in the manner described by Solomon et al. (1989). With this residual spectrum, using least square technique, column abundance of NO₃ had been calculated.

Results and discussions

Fig. 1 shows an example of the solar spectra taken during morning and noontime. These spectra contains Fraunhofer structure and absorption due to atmospheric species. Morning spectrum has been divided by the noon time spectrum to remove the Fraunhofer component

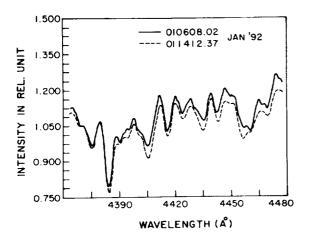


Figure 1. Zenith sky spectrum at 0802 hrs and 1237 hrs IST in the month of January.

present in it.

Fig. 2 shows an example of the ratio of the to noontime spectra. In this spectrum, Fraunhofer component should have been absent. But, in some cases the fluctuations in the ratio spectrum still remain. This is the Ring component. To remove this Ring contribution, we have taken the running average of ratio spectrum. The ratio spectrum thus obtained has the contribution of aerosols and Rayleigh To remove this component we have subtracted the ratio spectrum from a straight line parallel to the slope of the ratio spectrum. From the resultant residual spectrum, the slant column density of NO₂ and O₃ have been calculated using the method given by Harrison (1979). It may be mentioned here that in this method, only those wavelengths have been selected which have minimum contribution of Ring component. To calibrate the instrument, NO, spectrum has been scanned using artifical light source in 436 to 448 nm region, which shows good correlation with the residual spectrum obtained from atmosphere.

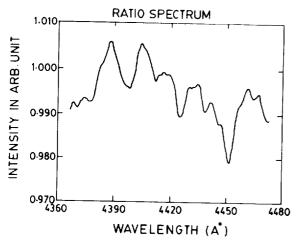


Figure 2. Ratio of twilight time and noon time spectra.

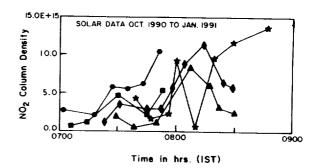


Figure 3. Increase of NO₂ vertical column density after sunrise.

Fig. 3 shows the vertical column density of NO_2 versus time for winter 1990-91. An increase in the value of NO_2 as the sun rises is clearly discernible. This is due to the formation of NO_2 by the photodissociation of N_2O_5 after sunrise.

Fig. 4 shows the plot of morning and evening vertical column abundances obtained during the period December 1990 to March 1992. After September 1991, the values of NO₂ appear to be slightly higher than those obtained in the earlier months. It may be mentioned here that Mt.Pinatubo erupted in June 1991 and it disturbed the aerosol distribution of the earth's atmosphere. After volcanic eruption, no alteration was made in the instrumental set-up. The interference due to additional aerosols on the retrieval process of NO, is minimum, as explained by Harrison (1979). We do not understand why their increase in NO₃. This could be due to the increase in value of albedo below acrosols layer which causes increase in the production rate of NO₂ compare to that before eruption period. The volcanic effect was more pronounced in the sunset period data compared to the sunrise period data. Contrary to our values, NO₂ column density observed by

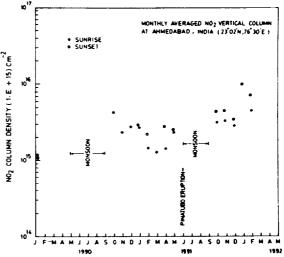


Figure 4. A plot of NO₂ vertical column density over the years.

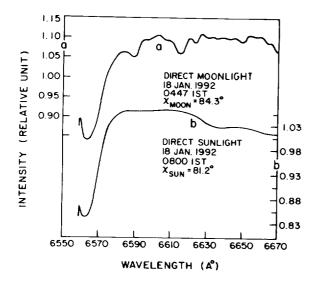


Figure 5. Direct solar and lunar spectra for large zenith angle.

similar technique at New Zealand (45°S, 170°E) shows decreasing trend after Pinatubo volcanic eruption. Before June 1991, the monthly vertical column abundance of NO₂ has been found to be in the range of 1-5 $\times 10^{15}$ cm². From September 1991 (in the post eruption period of Mt. Pinatubo) NO₂ values have been seen to increase, in February 1992, the NO₂ values are $\sim 10^{16}$ cm².

 $\sim 10^{16}$ cm $^2.$ Fig. 5a shows the background direct solar spectrum and Fig. 5b shows the observed direct lunar spectrum between 655 and 667 nm. Except $\rm H_{\alpha}$ at 656.3 nm, there is very little solar Fraunhofer structure in the

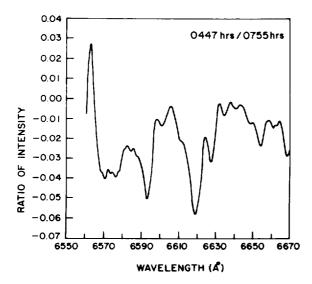


Figure 6. Residual absorption spectrum for 655 to 667 nm.

wavelength region used, thus the absorption feature due to NO_3 and H_2O can clearly be seen in the raw spectrum. After taking the ratio of observed to the background spectrum, ratio spectrum has been obtained. Since aerosol and Rayleigh scattering is also present in this wavelength region (i.e., 655 to 667 nm), a slope in the ratio spectrum has been observed. Subtracting the ratio spectrum from a straight line parallel to the slope of the ratio spectrum, residual absorption spectrum has been observed.

Fig. 6 shows the residual absorption spectrum obtained in the region 655 to 667 nm region. This spectrum contains absorption due to NO₃ and water vapour. NO₃ has maximum absorption at 662 nm which can clearly be seen in the residual spectrum as a prominent dip. Water vapour has also absorption at 659 nm region, which can be seen as a less pronounced dip compared to the NO₃. Besides this absorption feature, the Balmer-alpha feature is also seen in the residual absorption spectrum at 656.3 nm region. To verify the absorption feature of NO₃, observations have also been taken in the 617 to 629 nm region in which NO₃ has maximum absorption at 623 nm.

Fig. 7 shows the slant column abundance of NO₃ vs. lunar zenith angle observed on different nights. is to be noted that the slant column density of NO decreases as the lunar zenith angle increases. This is due to the modulation of aureole scattered flux at higher lunar zenith angle. The higher values of NO3 observed over Ahmedabad could be due to the tropospheric contribution from the industrial production, biomass burning in the tropics and NO transported from Kuwaity oil fires. Interestingly, NO₂ values measured simultaneously are also higher than the earlier measurements.

Fig. 8 shows the vertical column density of NO₃ against time observed on different nights. It is to be seen from this figure that after sunset NO₃

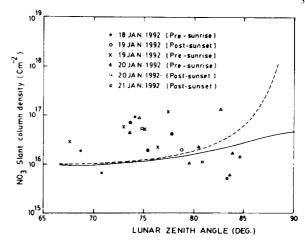


Figure 7. Slant column abundance of NO₃ vs lunar zenith angle.

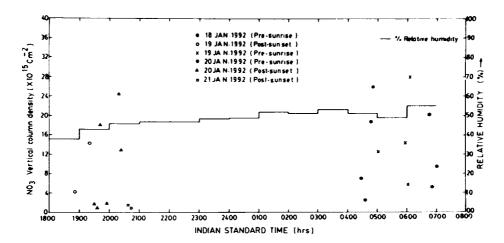


Figure 8. Vertical column abundance of NO3 against time, Relative humidity is shown by continous line.

This is in agreement with the results of Platt et al. (1981). Few observations have also been taken before sunrise, where NO, shows a decrease in column density as one approaches sunrise. relative humidity during the night has also been plotted in Fig. 8 which shows a positive correlation with NO₃ as observed by Platt et al. (1981). Our measured values of NO₂ lie between 1 and 20 x 10¹⁵ cm⁻².

Conclusion

In this paper we have presented results of NO₂ and NO, measurements by a ground-based visible absorption spectroscopy technique. Measurements have been made latitude northern hemispheric at a low Ahmedabad (23°N, 76°E), India. Several interesting The monthly vertical results have been obtained. The monthly vertical column abundance of NO₂ has been found to be in the range of 1-5x10¹⁵ cm². An increase in the value of NO₂ was found after Pinatubo volcanic eruption. values have also been observed during the night time. Its values varies from 1 to $20 \times 10^{15} \text{cm}^2$. This suggest that tropospheric contribution of NO3 is very high at our observation site.

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